

CHEMICAL CHANGES OF PUMPKIN SEED OILS AND THE IMPACT ON LIPID STABILITY DURING THERMAL TREATMENT: STUDY BY FTIR - SPECTROSCOPY

Fatos Rexhepi¹, Aziz Behrami¹, Cristina Samaniego-Sánchez², Maksim Rebezov^{3,4}, Mohammad Ali Shariati⁴, Artur Bastian da Silva⁵, Sávio Leandro Bertoli⁵ and Carolina Krebs de Souza^{5*}

Address(es):

¹ Faculty of Food Technology, University of Mitrovica "Isa Boletini" Ukshin Kovačica, 40000, Mitrovica, Kosovo.

² Department of Nutrition and Bromatology, Pharmacy Faculty UGR, 10871, Granada, Spain.

³ Department of Scientific Research, V. M. Gorbатов Federal Research Center for Food Systems, 109316 Moscow, Russian Federation.

⁴ Department of Scientific Research, K.G. Razumovsky Moscow State University of technologies and management (The First Cossack University), 73, Zemlyanoy Val St., Moscow, 109004, Russian Federation.

⁵ Department of Chemical Engineering, University of Blumenau, 89030-000, Blumenau, Santa Catarina, Brazil.

*Corresponding author: carolinakrebs@furb.br; shariatymohammadali@gmail.com

<https://doi.org/10.55251/jmbfs.5839>

ARTICLE INFO

Received 16. 11. 2021

Revised 10. 2. 2022

Accepted 15. 2. 2022

Published 1. 6. 2022

Regular article

OPEN ACCESS

ABSTRACT

Oxidative stability and fatty acid composition of the pumpkin seed oil (PSO) were studied, during roasting process from 80 to 150 °C and oil heat treatment (220 °C), by FTIR (Fourier - transform infrared) - Spectroscopy and GC/MS (Gas chromatography - mass spectrometry). The physicochemical parameters (density and iodine value) showed a significant increase ($p < 0.05$) during roasting, with a maximum value observed in the oil roasted at temperatures from 90 to 110 °C. The relative contents of polyunsaturated fatty acids (PUFAs) decreased to 87.7%, and saturated fatty acids (SFAs) increased to 112.5% at unroasted sample, after thermal treatment of oil samples. Especially at temperature 90 and 110 °C, oil samples have demonstrated that relative contents of PUFAs decreased to 95 % and SFAs increased to 101.6%. FTIR Spectroscopy provide to understand mechanism of chemical changes of seed during roasting process and, as result, which of obtained compounds are responsible for increasing thermal stability of oil lipids. In this study, it was observed that the best temperature for roasting pumpkin seeds is 110 °C, since at this point, both oxidation and Maillard reactions give rise to compounds with maximum antioxidant effect (lipid stability).

Keywords: Pumpkin seed oil; Ratio frequency; Maillard reaction; FTIR-Spectroscopy; GC/MS

INTRODUCTION

Pumpkin seed oil (PSO) from varieties Cucurbita Pepo L. pumpkin seed without shell, usually it used as a flavoured salad oil, is originated especially from Styria region in Austria and other European countries (Fruhirth and Hermetter, 2007). During roasting of pumpkin seeds, variations in chemical composition are observed relation to the unroasted flavor, especially the nutty flavor, whose properties obtained are from oxidized compounds, identified after the roasting process (Murkovič *et al.*, 2004; Siegmund and Murkovič, 2004).

Cooking is a necessary process before the consumption of many vegetables. It allows the improvement of sensory, nutritional and safety features, as well as modifies the phytochemical digestibility (Samaniego-Sánchez *et al.*, 2021). Researches highlights the importance of pre-roasting treatments to improve the quality of oil, attributing greater antioxidant effect (Nguyen, 2020) and higher lipid stability (Ali *et al.*, 2017; Aktaş *et al.*, 2018). Roasting process is crucial for obtaining necessary chemically changes of pumpkin seed oil, with best temperatures from 90 to 130 °C, shows maximum anti-radical activity at 110 °C (Potočnik *et al.*, 2018). For formation of responsible compounds for nutty roasted aroma, temperatures above 100 °C are applied. These compounds can be originated from of three reactions, which can occur during roasting process of pumpkin seed such as Strecker degradation, lipid peroxidation and Maillard reaction (Siegmund & Murkovič, 2004).

During roasting process of pumpkin seed, most of aldehydes compounds are product of strecker degradation and lipid peroxidation, which occur around 100 °C. Furan derivative compounds can be a product of lipid peroxidation and most of pyrazines can be of Maillard Reaction (Siegmund & Murkovič, 2004).

Roasting process of pumpkin seed results in chemical transformations, producing the mixture of different compounds that can play role of protective agents against lipid oxidation or their degradation (Potočnik *et al.*, 2018; Namiki *et al.*, 1988), which ensures higher antioxidant activity or stability of roasted pumpkin seed oil. Challenge of this research is to use FTIR-Spectroscopy, GC/MS and some physicochemical parameters to obtain more information about chemical reactions during roasting, as well as monitoring the stability of pumpkin seed oil, by

determining exactly at which roasting temperature the oil has the lowest level lipid oxidation or degradation.

There is several information in the literature about chemical changes in pumpkin seed during roasting, including the three type of reactions which occur during this process and their correlation with these chemical changes (oil stability and antioxidant capacity) during roasting temperature, where the chemical transformations are at their maximum activity. However, there is still unknown if all three reactions occur simultaneously in each roasting stage, and which would be the main reaction.

Nevertheless, detailed investigation of stability of roasted PSO, which determine responsible components for this antioxidant effect, will not be complete if are not involve the chemical changes in seed during roasting and determine the main reaction which occurs at every temperature, to then find interconnection between chemical composition and oil stability.

Therefore, the first aim of this study is to determine in which temperature the main reaction occurs (Strecker degradation, Maillard reaction, Lipid peroxidation or some of their correlation) using FTIR Spectroscopy and compare, by analysis of physicochemical parameters and GC/MS, the oil samples. The second aim is to verify the stability of unroasted and roasted PSO, at different temperatures, to determine in which temperature the oil shows higher stability.

MATERIAL AND METHODS

Material and chemicals

Pumpkin seed samples (Cucurbita pepo L) were obtained from Peja city (Kosovo) in 2019. Reagents used were iodobromine (IBr), glacial acetic acid (CH₃COOH), potassium iodide (KI), sodium thiosulphate (Na₂S₂O₃), n-hexane (CH₃(CH₂)₄CH₃), sodium methoxide (CH₃ONa) - FAME standard purity (≥ 99 %). All chemical reagents were purchased from Sigma-Aldrich, Germany.

Roasting conditions and oil extraction

Pumpkin seed were roasted in electric oven at 80 °C, 90 °C, 110 °C, 130 °C and 150 °C (Potocnik et al., 2018). The cold-pressed extraction of pumpkin seed oil, unroasted and roasted, was performed using a screw press (Koçmaksan, KMS10, Izmir, Turkey), according to Nederal et al. (2014).

Thermal oxidation - Thermal treatment process

Obtained oil from unroasted and roasted seeds samples (100 g), was inserted into 200 mL beakers, placed in an electric oven and thermally treated at 220 °C (10 min).

Density measurements

Densities of all oil samples, before roasting of their seeds and after thermally treated, were measured by an R.D bottle with a capacity of 5 mL (Zahir et al., 2017).

Iodine value (IV) measurements

A known weight of the oil sample was treated with iodobromine (IBr), in solution media of glacial acetic. Unreacted iodobromine, then react with potassium iodide, which converts it to iodine, which will be determined by titration with standard fresh prepared solution of sodium thiosulphate. Calculation of iodine value (equation 1):

$$IV = (b - v) * C * 126.9 * 100/v * 1000 \quad (\text{Eq. 1})$$

Where: *b* is the volume of thiosulphate used for blank sample, *v* is the volume of thiosulphate for real sample, *C* is the concentration of thiosulphate prepared solution, *w* is the weight of the real oil sample and 126.9 is the molecular weight of iodine (Singh, Gupta & Bajpai, 1981).

Spectral analysis

The results of thermal stability of the oil, obtained before and after roasting of the seeds, were recorded using instrument equipped with high sensitivity deuterated triglyceride sulphate (DTGS) – by FTIR detector (IRAffinity-1S Shimadzu Corporation, Kyoto, Japan). Samples were deposited in a CaF₂ transparent cell, spectroscopy running between 1000 - 4000 cm⁻¹, with 32 scans, resolution of 4 cm⁻¹ and recorded spot spacing of approximately 1.9 cm⁻¹. Peak intensity was calculated by FTIR IR Solution software.

Fatty acid composition

Fatty acid methyl esters (FAMES) were prepared, based on standard method 5509 (ISO) 1978 with minor modification described as follows: Weighted mass of 25 mg of oil samples were dissolved in 10 mL hexane and shaken. Then, 1 mL of sodium methoxide dissolved in methanol, with concentration 5.4 mol dm⁻³, were added in each of oil samples and mixed (60 seconds). After phase separation, from each sample, were collected supernatant for GC/MS analysis. The GC/MS used for FAME analysis was perform using a 7890A gas chromatograph equipped with auto sampler model 7693 and a 5975C MS detector (Agilent, Santa Clara, CA, USA). Used capillary column was DB-23 (50% bonded ciano polysiloxane 60 m x 0.25 mm i.e. x 0.25 μm stationary phase) and a helium as a carries gas. The split ratio mode of the sample was 1/25 and the injection volume was 3 μL for each analyzed sample. The temperatures of the GC system were the following: injector temperature 250 °C; transfer line temperature 280 °C; oven temperature program: 50 °C (2 min) –30 °C/min – 200 °C–20 °C/min–230 °C (15 min). Identifying and quantifying every peaks has been carried out in comparison with peaks of standard FAME, on their retention times, with those of external standards.

Statistical analysis

Statistical analyses were performed using the software Statistica (version 7.0, StatSoft Inc., Oklahoma, USA) (Hoffmann et al., 2021a,b,c). Normal distribution and variance homogeneity had been previously tested (Shapiro-Wilk). The experimental data were analysed by ANOVA, with the mean comparison (Tukey's test). The difference was considered significant if *p* < 0.05. All measurements were performed in triplicate and the results were reported as the mean ± one standard deviation.

RESULTS AND DISCUSSION

Pumpkin seed samples were wet salted and roasted at different roasting temperatures (80, 100, 120, 130 and 150 °C) (Potocnik et al., 2018). The oil samples of unroasted and roasted seeds, were submitted to analysis of density, iodine value, FTIR spectroscopy, Chromatography (FAME) and oxidative stability.

Density measurements

Vegetable oils are mixture of different triglycerides (TGs) and their density depends from nature of TGs, as well as of other compounds, which could be present in oils of different origins (Ramirez-Anaya et al., 2019; Ramirez-Anaya et al., 2015).

Density decreased linearly after increasing temperature due to the thermal expansion but continuously heating of oil, after its repeated thermal treatment, in general increases its value (Hoffmann et al., 2018; Kalogianni et al., 2011). During oil heat treatment, transformations of triglycerides occur, such as oxidation or polymerization. Density decreases with increasing unsaturation level, and the opposite occurs with increasing saturation and polymerization level (Kim et al., 2010).

Oxidised compounds originating, from those reactions, such as aldehydes, ketones, hydroperoxides or other unknown polymer compounds, may have higher or lower molecular weight compare to TGs, whose chemical changes will have impact in density changes. Paul & Mittal (1996) reported about occurring polymerization during heating of the oil and observed that the density increased. The oxidation and saturation of TG's is the first stage before polymerization process in edible oils (Choe & Min, 2006). A completely different chemical transformation occurs in pumpkin seeds during their roasting, because there is a mixture of compounds such as proteins, TGs, and natural constituents, which results in several chemical interactions.

Figure 1 shows the behaviour of the density of the oil, obtained from cold pressing of the pumpkin seeds, before and after roasting at different temperatures and after oil thermal treatment. At first sample (0), unroasted seeds, density is lower (0.90 g.cm⁻³) and gradually increased to maximum level (0.96 g.cm⁻³) at seed roasting temperature at 90 °C. From this density changes, it can confirm that sample roasted at 90°C contain higher level of oxidised or saturated compounds compare to, unroasted sample and roasted at 80 °C (Kim et al., 2010). From this point of view, during roasting up to 90°C, lipid peroxidation occurs, which results in lipid compounds of higher molecular weight in the first stage, which may subsequently decompose and produce other oxidised compounds with lower molecular weight (aldehydes, ketones, alcohol). This would increase the concentration of saturated fatty acids and consequently, the density of the oil (Barthel & Grosch, 1974; Zahir et al., 2017).

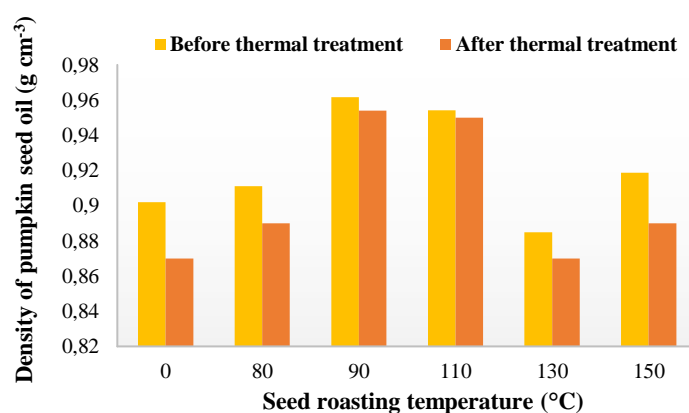


Figure 1 Pumpkin seed oil density, before (0) and after roasted (80, 90, 110, 130 and 150 °C), and before and after thermal treatment at 220 °C.

The oil samples of roasted seeds at 90 and 110 °C, had presented similar densities (0.95 g.cm⁻³). However, there is a reduction in the density of the samples roasted at 130 (0.88 g.cm⁻³) and 150 °C (0.91 g.cm⁻³). This could be explained with the increase of smaller compounds in the mixture, which indicates the formation unknown compounds from complex reaction, which occur in pumpkin seeds during roasting at higher temperature. Only the sample of seed oil roasted at 110°C showed lesser difference in density (*p* > 0.05), before and after thermal treatment. All the other samples showed a higher density after thermal treatment, possibly due to increased saturation level during thermal treatment or oxidation of their lipids.

Iodine value (IV) measurements

In Figure 2 it is presented the iodine value of oil samples, before seeds roasting at different temperatures and after seed thermal treatment, for determination of unsaturation level of lipids. The unroasted sample (0) and roasted samples at 80, 90 and 110 °C (*p* > 0.05), present similarity in iodine value, while the samples

roasted at higher temperature (130 and 150 °C) showed a significant decrease of iodine value (110 and 98, respectively).

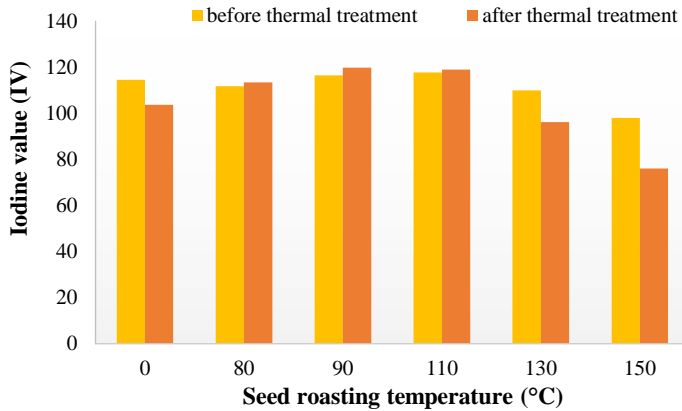


Figure 2 Pumpkin seed oil iodine value, before (0) and after roasted (80, 90, 110, 130 and 150 °C), and before and after thermal treatment at 220 °C.

The iodine value, of the same samples, was analyzed before and after thermal treatment, to check the stability of the oil. Before thermal treatment, the oil unroasted seed sample presents approximately the same iodine value ($p > 0.05$) as that of the samples roasted at 80, 90 and 110 °C (115 ± 2.7). However, unroasted samples, after thermal treatment, decreases iodine value, except the samples roasted at 80, 90 and 110 °C, especially at 110 °C, whose iodine value, after thermal treatment, increases slightly. This means that the oil obtained from pumpkin seed at this roasting temperature, is more stable compared to the oil obtained from the unroasted sample (Nederal et al., 2012).

Spectral analysis

FTIR spectra of oil samples, obtained from pumpkin seed roasted process, was used as indicator for lipid oxidation and as a basis for calculating the peak ratio 3007/2854. That methodology is used to characterize edible oils and fats since they differ in the intensity and exact frequency at which the transmittance or absorbance band maximum occurs, according to the composition and nature of the sample (Guillen and Cabo, 2000). Peak 3007 cm^{-1} (Figure 3) present cis double bond (=C-H) and normally during oxidation of this double bond, this will be converted in single methylene bond (2854 cm^{-1}). That mean if we use ratio of these, the same could be interesting indicator in the case of ratio decreasing, which could be understands as the occur lipid peroxidation (Guillen & Cabo, 2000).

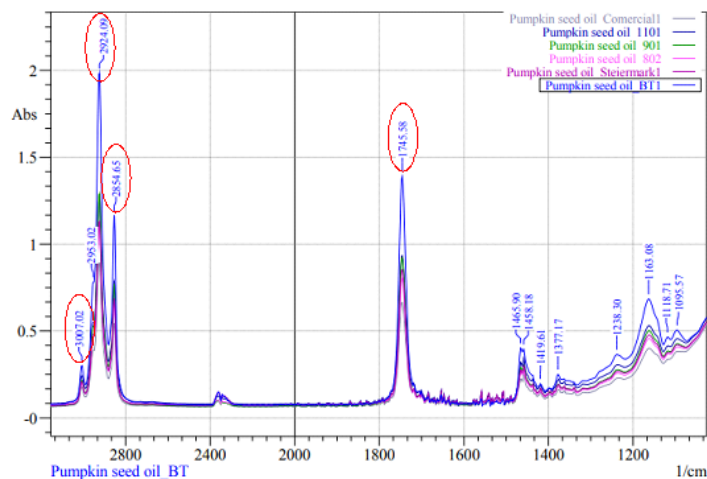


Figure 3 FTIR Spectra of pumpkin seed oil.

If double bonds are part of triglyceride compounds in that case could be used ratio 3007/1745 (Figure 4), because peak 1745 cm^{-1} it is a vibration of carbonyl group (C=O) ester functional group, representative as triglyceride compounds (Rexhepi et al., 2019; Guillen & Cabo, 2000). This ratio could be used as a second confirmation indicator for lipid peroxidation because vegetable oils composition is more than 95 % triglyceride and usually most of double bonds are part of triglyceride (Hoffmann et al., 2018). In Figure 4 are presented changes of both ratios, which are in same tendency of behaviour, and again would confirm that monitored double bonds are part of triglyceride composition and chemical transformation occur inside triglyceride molecules.

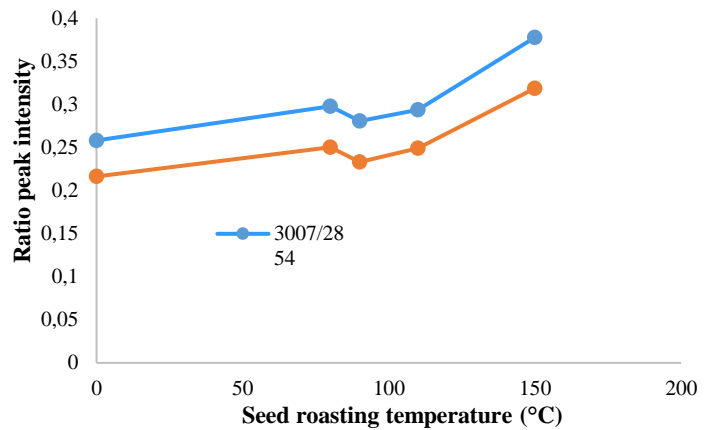


Figure 4 Behavior of the ratio intensity, of oil obtained from pumpkin seed unroasted (0) and roasted, at 80, 90, 110, 130 and 150 °C.

First monitored point is the unroasted sample and second point is the sample after roasting at 80 °C, which has increased ratio compare to the previous sample. This could be explained with the evaporation of volatile compounds that, in general, indicates that the level of unsaturated composition has increased. Next monitored sample is the roasted at 90 °C, which present lower level of this ratio, which confirm of lipid peroxidation occurred during roasted at this temperature. In this process, slightly increasing proportions of lipid peroxidation were observed, in addition to the percentage increase of the unsaturated composition. Increasing of unsaturated level from roasted sample at 110 °C was observed as a rapidly changes. These unexpected changes could be explained by the occurrence of lipid peroxidation at roasting temperature from 80 to 100 °C and their products could have other unsaturated chemical bonds which contribute in increasing in presented ratio.

Figure 5 present changes of three intensity bands, which are responsible for Maillard reaction, where occur same time, increased carbonyl group of triglyceride 1745 cm^{-1} and methylene group in two band symmetric and asymmetric stretching vibration, 2924 and 2853 cm^{-1} (Calabrò & Magazù, 2012). Intensity main increasing, for three bands same time, can be seen in third sample after roasting (90 °C). This fact stands out because only one stage occurs after lipid peroxidation. From this point of view Maillard reaction occur immediately after the end of lipid peroxidation, which as a primary reaction it seems to induce induces the Maillard reaction.

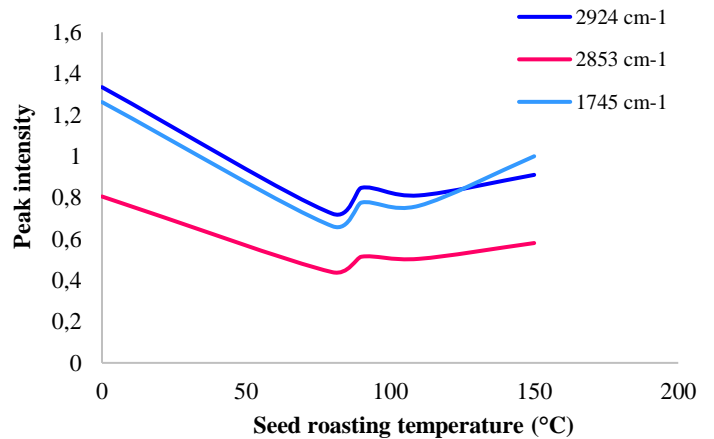


Figure 5 Changes of three vibrational band responsible for Maillard reaction, of oil obtained from pumpkin seed unroasted (0) and roasted at 80, 90, 110, 130 and 150 °C.

As conclusion for this mechanism, Maillard reaction start after the end of lipid peroxidation, that mean product obtained from lipid peroxidation play a role as a precursor for initiate the Maillard reaction. In general this indicate the interaction between carbonyl group of different sugars isomers or same group from other compounds, and free amino group from protein or individual aminoacids (Maillard, 1912).

Maillard products are responsible for nutty roasted aroma and based on the mechanism they are not possible to be obtained if not happened first lipid peroxidation and their product to be as a precursor for Maillard reaction. Maillard reaction it is still active after sample roasting at 110 °C, but not to the same degree as the previous sample (90 °C). After roasting temperature at 130 °C, they no longer occur. Strecker degradation it is another possibility for aminoacids converts in aldehyde, but not under interaction with sugars like Maillard reaction.

Strecker reaction involve oxidative deamination and decarboxylation of aminoacids in the presence of dicarbonyl compounds, whose formed components are aldehyde with one less carbon atom than the original amino acid (Whitfiel, 1992). Resulting compounds from Strecker degradation have low odour and taste thresholds. For this reason, they are important components for the final aroma of the food, drink or in this case vegetable oil. Anyway still is unknown if during pumpkin seed roasting process occurring Strecker degradation or not because any characteristic indicator isn't selective just for product of Strecker degradation applied in complex food samples such as pumpkin seed. Aldehyde products could be as resulting compounds from Strecker degradation and from Maillard reactions. From this point of view, occurring of Strecker degradation during pumpkin seed roasting is a topic, whose occurrence, has been little detailed and proven by the scientific community.

Considering the possibility that this reaction also occurs during seed roasting, it would be important to understand which reaction predominates, to explain these mechanisms and the possibility that Strecker degradation interacts with other chemical transformations.

To monitor thermal oxidative stability level, of PSO, for each roasted sample at different temperature and compare with same samples after thermal treatment, was used FTIR band of frequency ratio 3007/2854 cm⁻¹, which is used indicator for oxidation level of lipids. As can be seen in Figure 6, in general all heat-treated samples have higher oxidation degree, this can be explained probably because during their thermal treatment evaporate most of volatile compounds which change their oxidation level and as a result increases the concentration of unsaturated compounds.

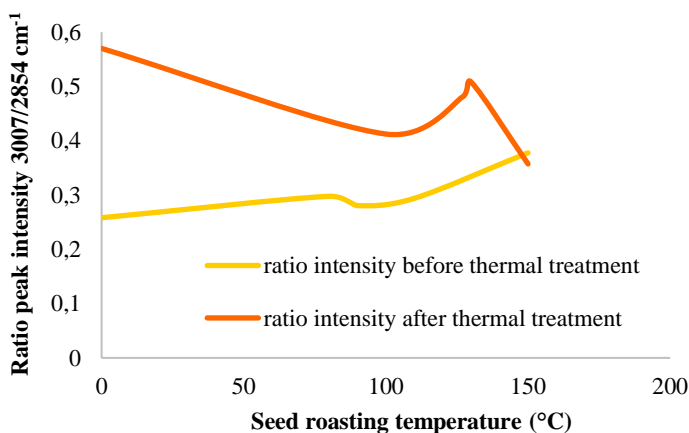


Figure 6 Ratio of intensity 3007/2854 cm⁻¹ for pumpkin seed oil, before and after thermal treatment (oxidative stability).

Roasted sample has low difference ratio, before and after thermal treatment, compare with unroasted, which has higher ratio difference before and after thermal treatment. This can be explain if roasted sample, at temperature from 80 to 130 °C, contain compounds with high antioxidant effect, which protect lipids from oxidation. Sample roasted at 150 °C rapidly decreases ratio of frequencies, and this mean decrease of unsaturated level of lipids is because occur oxidation of double bonds in lipid structure. In this case, oxidised compounds are insufficient to protect lipid oxidation, because they could be decomposed or they changed chemically. Another FTIR important band is frequency from 3000-3020 cm⁻¹, characteristic from double bonds (=C-H), which indicates of triglycerides unsaturation level (lipid oxidation) (Vlachos et al., 2006).

Figure 7 presents the changes of peak area, referring to the oxidative stability of the pumpkin seed oil samples, unroasted and roasted, before and after thermal treatment (220 °C).

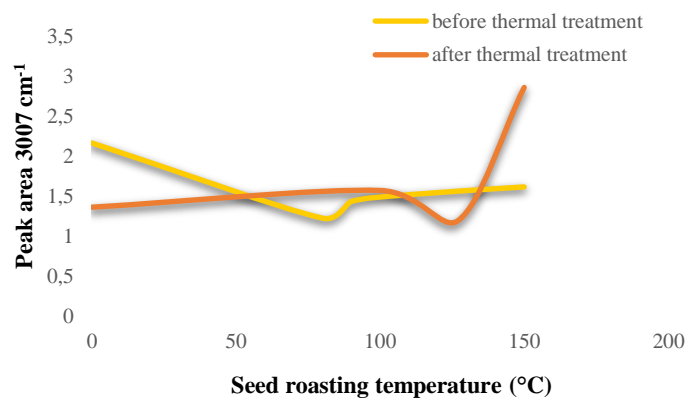


Figure 7 Peak area of band 3007 cm⁻¹ for pumpkin seed oil before and after thermal treatment (oxidative stability).

Only unroasted sample presented higher difference before (1.36) and after (2.17) thermal treatment, which confirm the lipid instability of this sample compare to roasts. Less variation is observed in the values, before and after thermal treatment, of samples roasted from 90°C to 110°C, indicating greater lipid stability in this temperature range, compared to samples roasted above 110°C (Ibsch et al., 2019). It is possible that at higher temperatures unknown reactions start to occur during roasting of the seeds. Enzyme inactivation, described as responsible for the degradation of oils obtained after heat treatment of seeds, is a factor that could explain the increased lipid stability of roasted pumpkin seed oil (Durmaz & Gökmen, 2010; Ibsch et al., 2019; Schmitz et al., 2021). However, based on this research, enzyme inactivation would not explain the oxidative stability being higher in samples roasted from 80 to 130°C, since from this temperature the stability of the oil decreases. For this reason, it is believed that roasted PSO, presents products that contribute to oxidative stability, with antioxidant effect, up to the roasting temperature from 80 to 130 °C (especially 110 °C).

Researchers found higher total polyphenols contents, in oil obtained from roasted pumpkin seed, and reported formation of different antioxidant compounds (Nawirska-Olszańska et al., 2013), probably, from Maillard reaction, which most of them can have protective activity on lipid oxidation (Nederal et al., 2012; Veronezi & Jorge, 2012). Murković & Pfannhauser (2020) also investigated about stability of pumpkin seed oil after roasting, and they evidenced the influence of the ratio of linoleic and oleic acid in their lipids. These researchers emphasize the importance of studies on the oxidative behaviour after roasting of pumpkin seeds, as many antioxidant compounds are present after the roasting process.

FTIR – Spectroscopy, as a sensitive tool for monitoring lipid peroxidation, was applied using ratio intensity of different frequencies, to verify chemical correlations (Rexhepi et al., 2019; Guillen & Cabo, 2000). This tool has also been used to detect the Maillard Reaction by monitoring the intensity of the carbonyl group of triglycerides and methylene group of aliphatic hydrocarbons. When both intensity frequencies increase, it can be attributed to the partial occurrence of the Maillard reaction (Calabrò & Magazù, 2012).

Fatty acid composition

Fatty acids levels are showed in Table 1. The main fatty acids in the pumpkin seed oil samples are linoleic acid, oleic acid, palmitic acid, stearic acid. The average of fatty acid content in analyzed samples varies around 99 %, which is higher than those published by Nederal et al. (2012).

Table 1 Fatty acid composition (%) of unroasted and roasted pumpkin seed oils, before thermal treatment at 220 °C.

Fatty acids	Roasting temperature (°C)					
	unroasted	80	90	110	130	150
Lauric (C12:0)	0.02±0.01 ^a	0.02±0.01 ^a	0.02±0.01 ^a	0.02±0.01 ^a	0.02±0.01 ^a	0.02±0.01 ^a
Myristic (C14:0)	0.10±0.01 ^a	0.10±0.01 ^a	0.10±0.01 ^a	0.09±0.01 ^a	0.09±0.01 ^a	0.10±0.01 ^a
Palmitic (C16:0)	11.16±0.01 ^a	11.17±0.01 ^a	11.23±0.01 ^a	11.28±0.01 ^a	11.53±0.01 ^a	11.48±0.01 ^a
Palmitoleic (C16:1)	0.12±0.02 ^b	0.11±0.02 ^b	0.11±0.02 ^b	0.11±0.02 ^b	0.15±0.01 ^a	0.12±0.02 ^{ab}
Heptadecanoic (C17:0)	0.07±0.02 ^a	0.06±0.01 ^a	0.06±0.001 ^a	0.06±0.01 ^a	0.07±0.01 ^a	0.07±0.01 ^a
Stearic (C18:0)	9.69±0.01 ^a	9.73±0.01 ^a	9.73±0.01 ^a	9.72±0.01 ^a	9.72±0.01 ^a	10.06±0.05 ^a
Oleic (C18:1)	36.54±0.03 ^a	36.75±0.03 ^a	36.81±0.03 ^a	36.50±0.03 ^a	37.31±0.03 ^a	36.19±0.03 ^a
Linoleic (C18:2)	41.33±0.04	41.27±0.04	41.27±0.04	41.17±0.04 ^a	40.36±0.04 ^a	41.19±0.05 ^a
Linolenic (C18:3)	0.13±0.01 ^b	0.13±0.01 ^b	0.13±0.01 ^b	0.13±0.01 ^b	0.18±0.01 ^a	0.17±0.01 ^a
Arachidic (C20:0)	0.53±0.01 ^b	0.52±0.01 ^b	0.52±0.01 ^b	0.50±0.01 ^b	0.56±0.01 ^a	0.58±0.01 ^a
ΣSaturated	21.87 ^a	21.66 ^a	21.66 ^a	21.546 ^a	21.98 ^a	22.31 ^a
ΣMonounsaturated	36.66 ^a	36.92 ^a	36.92 ^a	36.60 ^a	37.46 ^a	36.31 ^a
ΣPolyunsaturated	41.46 ^a	41.40 ^a	41.40 ^a	41.84 ^a	40.54 ^a	40.36 ^a
P/S ratio	1.89 ^a	1.90 ^a	1.91 ^a	1.94 ^a	1.84 ^a	1.85 ^a

Means ± standard deviation fatty acid composition of pumpkin seed oils.

Different lowercase letters on the same line indicate a significant difference at the 5% level by Tukey's test.

The fatty acid profile of vegetable oils influences its physical properties during treatment, especially levels of viscosity and heat transfer properties (Debnath, Vidyarthi & Singh, 2010). Usually during oil heating occur lipid oxidation, increasing the concentration of saturated fatty acids and decreasing of unsaturated fatty acids, which is probably a result from PUFA conversion in MUFA and SFA (Henna & Tan, 2009). Its degree of unsaturation however, modifies the stability during thermal treatment (Contreras-Gallegos et al., 2017).

An increase in the percentage of palmitic and stearic acid in the oil obtained after roasting the seeds is observed as the roasting temperature increases. Except the sample at 150 °C, whose palmitic acid does not maintain this trend. However, linoleic acid decreased its percentage as the roasting temperature increased, except in the sample roasted at 150°C.

Usually unsaturated fatty acids during oxidation are converted to saturated corresponding fatty acid but the percentage of oleic acid, on the other hand, did not correlate with the increase in roasting temperature, probably due to the different reactions that may occur at different temperatures. These results in the percentages of fatty acids, indicate that the expected oxidation did not occur, since other reactions may occur at different stages and temperatures of the roasting process (Lee, Kim, & Choe, 2007). This was also confirmed by the percentages total of saturated and unsaturated fatty acids that also showed unexpected variations. In general, the percentage SFA increased as the percentages PUFA and MUFA decreased, especially during roasting at higher temperatures. It is important to note that this usually happens at higher roasting temperatures, where completely different reactions probably occur than at lower temperatures. The P/S ratio is also important, as when it has decreased, it indicates that there has been lipid oxidation in the oil sample. The unexpected values of this parameter increase in the sample roasted at 110°C, which could be explained by lipid peroxidation reactions and Maillard reaction that occur especially in this temperature range. Probably these reactions are in the intermediate stage where polyunsaturated fatty acids are not converted into saturated fatty acids. According to Raczky et al. (2017), roasting of pumpkin seed at temperature from 100 to 130 °C, present higher oxidative stability of obtained oil, without any effect at the composition on their fatty acid.

CONCLUSION

For all the studied samples, before and after roasting, the obtained oil showed a change in density, which demonstrates the occurrence of oxidation and saturation especially in the samples whose seeds were roasted from 90 to 110 °C. The iodine value shows similar changes, which confirms that the oil did not change at the saturation level of the sample roasted at 110°C, reducing its values at higher temperature. Both the density and the iodine value, showed little variation after heat treatment at 220 °C, especially in the roasted seed oil samples at 90 and 110 °C, demonstrating greater stability in this temperature range. Based on FTIR measurements ratio intensities band and single band, lipid peroxidation occurs from roasting temperature from 80 to 90°C, and from 90 to 110 °C Maillard reaction occurs. Thus, lipid peroxidation compounds are precursors of Maillard reactions. The stability of the oil, in this temperature range, may be correlated to the origin of oxidized compounds that have antioxidant effect. Finally, the best temperature for roasting pumpkin seeds is 110 °C, since at this point, both oxidation and Maillard reactions give rise to compounds with maximum antioxidant effect (lipid stability).

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