

## BOX-BEHNKEN EXPERIMENTAL DESIGN MEDIATED OPTIMIZATION OF AQUEOUS METHYLPARATHION BIODEGRADATION BY *Pseudomonas aeruginosa* mpd STRAIN

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doi: 10.15414/jmbfs.2016.5.6.534-547

### ARTICLE INFO

Received 22. 2. 2015  
Revised 14. 11. 2015  
Accepted 24. 1. 2016  
Published 1. 6. 2016

Regular article



### ABSTRACT

Biotreatment of methylparathion was studied in aqueous mineral salts medium containing bacterial culture to demonstrate the potential of the novel strain of *Pseudomonas aeruginosa* mpd. A statistical Box–Behnken Design (BBD) of experiments was performed to evaluate the effects of individual operating variables and their interactions on the methylparathion removal with initial concentration of 1000 mg l<sup>-1</sup> as fixed input parameter. The temperature (X<sub>1</sub>), pH (X<sub>2</sub>), reaction time (X<sub>3</sub>) and agitation (X<sub>4</sub>) were used as design factors. The result was shown that experimental data fitted with the polynomial model. Analysis of variance showed a high coefficient of determination value 0.9. The optimum biodegradation of MP in terms MP removal (Y<sub>1</sub>), COD removal (Y<sub>2</sub>) and TOC removal (Y<sub>3</sub>) were found to be 95.2 %, 82 % and 61.2 % respectively. The maximum growth (Y<sub>4</sub>) was 2.18 optical density (OD). The optimum biodegradation correspond to the factors combination of middle level of X<sub>1</sub> (33 °C), X<sub>2</sub> (7.0), X<sub>4</sub> (150 rpm) and the highest level of X<sub>3</sub> (96h). MP removal and its residues were detected using spectral analysis. The study demonstrates the optimum MP biodegradation potential of this strain could use MP as the sole Carbon/Phosphate source. BBD confirmed to be dependable in developing the model, optimizing factors and analyzing interaction effects. Data from this study will be helpful in the design of small-scale field experiments and subsequently an in situ methylparathion biotreatment system for field application.

**Keywords:** Wastewater Biotreatment. Design Optimization. *Pseudomonas aeruginosa*. Biodegradation. *O,O*-dimethyl *-O*-4-nitrophenylphosphorothioate

### INTRODUCTION

Continuous and excessive use of organophosphorus (OP) compounds has led to the contamination of several ecosystems in different parts of the world (Cisar and Snyder, 2000; Tse et al., 2004). Thiophosphoric acid esters, such as parathion, methylparathion (MP) and tetrachlorvinphos, are hazardous pollutants and their accumulation in the environment is a recognized ecological threat (Kaloyanova and Tarkowski, 1981). Methods for their enhanced degradation are an urgent task of contemporary chemical technology and biotechnology. Its widespread use has caused environmental concern due to its frequent leakage into surface and ground waters. The drinking water directive (Council directive 98/83/EC) sets an allowed contaminant level of 0.1 mg/L for a single pesticide and 0.5 mg/L for the total sum of pesticides. Industries manufacturing pesticides release wastewater in water bodies or land. Although industries treat their wastewater by activated sludge process, no attention is paid to remove the specific pesticides or their metabolites which exert toxicity at very low concentrations. Therefore, there is a need for economically dependable methods for organophosphorus (OPs) detoxification from the environment. To date, bacterial transformations have been the main focus in research on organophosphate pesticide degradation. *Pseudomonas aeruginosa*, *Clavibacter michiganense* (Subhas and Dileep, 2003), *Arthrobacter atrocyaneus*, *Bacillus megaterium* and *Pseudomonas mendocina* (Bhadbhade et al., 2002), *Agrobacterium radiobacter* (Horne et al., 2002), and other *Pseudomonas* species (Ramanathan and Lalithakumari, 1999) have been reported to degrade OP in solutions and soils. Use of specific microorganism adapted to the pesticides, in treatment of industrial effluents is not in practice (Kanekar et al., 2004). Therefore, research should be concentrated to develop economical but effective microbial processes for the treatment of industrial effluents containing pesticides and take them to field. The aim of this research was to optimize the process variables for the biodegradation potential of the *Pseudomonas aeruginosa* mpd novel strain using response surface methodology (RSM).

### MATERIAL AND METHODS

#### Bacterial culture conditions

A potential bacterial strain (*Pseudomonas aeruginosa* mpd) was isolated from pesticide exposed agricultural soil. The initial enrichment cultures were established in a synthetic wastewater containing mineral salts medium amended with the methylparathion (Devithion™ 50% EC) as the sole source of carbon and energy. The concentration of methylparathion used was 0.1%, pH was adjusted using 1N NaOH and 1N HCl (ELICO - L1127, India). The methylparathion contaminated synthetic wastewater was neutral pH and the mean value of methylparathion (MP) content was 1000 mg/L, chemical oxygen demand (COD) was 41950 mg/L and total organic carbon (TOC) was 10459 mg/L. The synthetic wastewater containing higher concentration of methylparathion with maximum level of 1000 mg/L was used in the present study. Stock solution of pure methylparathion (98.5%) was prepared by dissolving 1g in 100 mL methanol, made up to 1000 mL of distilled water and was used as a reference for instrumental analysis. It was reported that the pesticide pollution due to wastewater released from formulating or manufacturing pesticide industry were up to 1000 mgL<sup>-1</sup> (Chiron et al., 1997). Therefore in this research, synthetic wastewater containing methylparathion with maximum concentration of 1000 mgL<sup>-1</sup> was used.

Organisms were subsequently grown on nutrient agar medium plates to obtain single colonies. A pure culture of methylparathion-degrading *Pseudomonas aeruginosa* was isolated by series of replating on MSM with methylparathion agar plates. Minimum inhibitory concentration (MIC) test with plate screening method was carried out to screen methylparathion resistant bacteria using methylparathion MSM with methylparathion agar plates. Based on the MIC test, the five potential bacterial cultures isolated were identified based on their morphological characters and biochemical tests as given in Bergey's Manual of Determinative Bacteriology (Holt et al., 2000). For degradation studies, *Pseudomonas aeruginosa* was inoculated into sterile shake-bottles containing 250

mL of MSM, 0.1% (w/v) methylparathion and incubated under aerobic conditions on a shaker (150 rpm) for 96 h. The other parameters, i.e., pH value, culture temperature, time and agitation, were part of the experimental design. All experiments were performed in triplicate, and the results are expressed as an average of three replicates.

**Optimization of methylparathion degrading condition by *Pseudomonas aeruginosa* mpd**

In order to study the effect of variables on the degradation of methylparathion by biotreatment using potential bacterial strain, the process variables include pH, temperature time and agitation were optimized. Experimental design was set using the variables such as pH, temperature, time and agitation. The synthetic wastewater which consists of mineral salts medium amended with the methylparathion was set at various temperature (25-40°C), pH (5-9), time (24-168 h) and agitation (120-180 rpm) for analysis. The concentration of methylparathion used was 1000 mgL<sup>-1</sup>. The pH was adjusted using 1N NaOH and 1N HCl with the help of pH meter (ELICO - L1127, India). During this process, estimation of various parameters such as residual methylparathion, COD removal, TOC removal and pH were analysed to measure the degradability of methylparathion. Response surface methodology (RSM) based on the Box-Behnken design of experiment was used to optimize these parameters and their interaction which significantly influenced methylparathion biodegradation.

**Box–Behnken Experimental Design (BBD) of methylparathion bioremoval using RSM**

A standard RSM design called Box-Behnken’s Design (BBD) for biotreatment process was adopted to study the influence of variables for the removal of

aqueous methylparathion. The method can reduce the number of experimental trials needed to evaluate multiple parameters and their interactions and for finding the most suitable condition and prediction of response (Box and Behnken, 1960; Myers and Montgomery, 2002). Among all the RSM designs, BBD requires fewer runs than the others, e.g., 29 runs for a 4-factor experimental design. By careful design and analysis of experiments, Box-Behnken design allows calculations of the response function at intermediate levels which were not experimentally studied and shows the direction if one wishes to change the input levels to determine the effects on the response (Hamed and Sakr, 2001, Martínez-Toledo and Rodríguez-Vázquez, 2011).

**Table 1** The levels of variables in Box-Behnken statistical experiment design

Variable	Name	Coded level		
		-1	0	+1
X <sub>1</sub> :A	Temperature (°C)	25	32.5	40
X <sub>2</sub> :B	pH	5	7	9
X <sub>3</sub> :C	Time (h)	24	96	168
X <sub>4</sub> :D	Agitation (rpm)	120	150	180

The relation between the code values and none code values were:

$$X_1 = (A - 32.5)/7.5, X_2 = (B - 7)/2, X_3 = (C - 96)/72, X_4 = (D - 150)/30.$$

**Table 2** Experimental design with coded and actual values

Run	Temp (°C)	pH	Time (h)	Agitation (rpm)	Temp (°C)	pH	Time (h)	Agitation (rpm)
Coded Values					Actual Values			
1	1	0	0	-1	40	7	96	120
2	0	1	1	0	32.5	9	168	150
3	0	0	-1	-1	32.5	7	24	120
4	-1	0	0	-1	25	7	96	120
5	1	0	1	0	40	7	168	150
6	1	1	0	0	40	9	96	150
7	-1	0	1	0	25	7	168	150
8	0	0	1	1	32.5	7	168	180
9	-1	0	-1	0	25	7	24	150
10	1	0	-1	0	40	7	24	150
11	0	-1	0	-1	32.5	5	96	120
12	0	0	1	-1	32.5	7	168	120
13	0	0	0	0	32.5	7	96	150
14	-1	0	0	1	25	7	96	180
15	0	1	-1	0	32.5	9	24	150
16	0	0	0	0	32.5	7	96	150
17	0	1	0	1	32.5	9	96	180
18	0	-1	-1	0	32.5	5	24	150
19	0	1	0	-1	32.5	9	96	120
20	0	-1	1	0	32.5	5	168	150
21	1	0	0	1	40	7	96	180
22	0	-1	0	1	32.5	5	96	180
23	0	0	0	0	32.5	7	96	150
24	0	0	0	0	32.5	7	96	150
25	-1	1	0	0	25	9	96	150
26	-1	-1	0	0	25	5	96	150
27	0	0	0	0	32.5	7	96	150
28	1	-1	0	0	40	5	96	150
29	0	0	-1	1	32.5	7	24	180

Response surface methodology (RSM) based on the BBD of experiment was used to optimize the variables and their interaction which significantly influenced methylparathion biodegradation by the individual strains of *Pseudomonas aeruginosa* mpd. A four-factor, three-level Box-Behnken design was used in the biotreatment process. The Box-Behnken design is an independent, rotatable quadratic design with no embedded factorial or fractional factorial points where the variable combinations are at the mid-points of the edges of the variable space and at the center. Among all statistical experiment designs, Box-Behnken design requires fewer runs than the others, e.g., 29 runs for a 4-factor experimental

design. The low, middle and high levels of each variable were designated as -1, 0, and +1 respectively, as given in Table 1. For this biotreatment process, the variables and their values in brackets were three levels include temperature (25-40°C), pH (5-9), time (24-168 h) and agitation (120-180 rpm), at constant methylparathion concentration 1000 mgL<sup>-1</sup>(0.1%). This also enabled the identification of significant effects of interactions for the batch studies. This also enabled the identification of significant effects of interactions for the batch studies. In system involving four significant independent variables X<sub>1</sub>, X<sub>2</sub>, X<sub>3</sub>, and X<sub>4</sub>, the mathematical relationship of the response of these variables can be



$$Y_2 = 82.0 - 0.757X_1 + 2.5X_2 + 1.33X_3 + 2.58X_4 - 13.25X_1^2 - 11.87X_2^2 - 15.62X_3^2 - 17.50X_4^2 - 1.50X_1X_2 + 1.0X_1X_3 - 4.25X_1X_4 + 1.0X_2X_3 + 1.0X_2X_4 + 2.50X_3X_4 \quad (2)$$

$$Y_3 = 61.2 - 1.257X_1 + 2.58X_2 + 0.92X_3 + 1.75X_4 - 15.98X_1^2 - 12.73X_2^2 - 15.47X_3^2 - 19.22X_4^2 - 0.25X_1X_2 + 2.25X_1X_3 - 4.75X_1X_4 + 4.25X_2X_3 + 2.25X_2X_4 + 1.25X_3X_4 \quad (3)$$

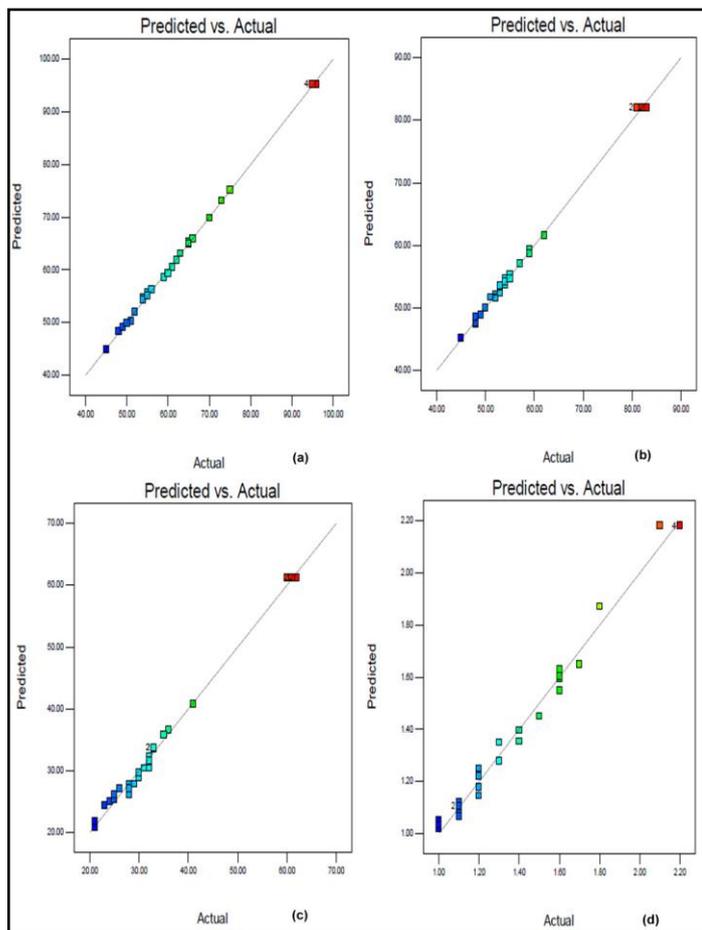
$$Y_4 = 2.18 - 0.075X_1 + 0.13X_2 + 0.13X_3 + 0.12X_4 - 0.38X_1^2 - 0.44X_2^2 - 0.38X_3^2 - 0.56X_4^2 - 0.025X_1X_2 - 0.025X_1X_3 - 0.17X_1X_4 + 0.25X_2X_3 + 0.13X_2X_4 + 0.15X_3X_4 \quad (4)$$

Where  $Y_1$  (% methylparathion removal),  $Y_2$  (% COD removal),  $Y_3$  (% TOC removal) and  $Y_4$  (bacterial growth in OD) is the predicted responses where as the  $X_1$  (temperature),  $X_2$  (initial pH),  $X_3$  (time) and  $X_4$  (agitation) are the coded variables.

**Table 3** The observed (experimental) values and model response (predicted) values obtained from combination of process variables

Run	$Y_1$		$Y_2$		$Y_3$		$Y_4$	
Run	MP Removal (%)		COD Removal (%)		TOC Removal (%)		Growth-MP <sup>+</sup> (OD)	
	EV	PV	EV	PV	EV	PV	EV	PV
1	75	75.17	52	52.17	28	27.75	1.2	2.352
2	62	61.84	59	59.34	41	40.75	1.8	1.871
3	55	55.63	48	47.47	24	25.09	1.2	2.277
4	51	50.35	45	45.17	21	20.75	1	2.152
5	70	69.88	55	54.71	32	31.67	1.5	1.451
6	63	63.13	57	57.13	33	33.57	1.4	1.396
7	52	52.06	54	54.21	30	29.67	1.7	1.651
8	62	61.81	55	55.29	32	30.43	1.6	2.759
9	49	49.06	53	53.55	32	32.33	1.3	1.351
10	50	49.88	50	50.05	25	25.33	1.2	1.251
11	65	60.38	48	48.55	28	27.17	1	2.181
12	73	73.13	45	45.13	23	24.43	1.1	2.227
13	95	95.2	81	82	60	61.2	2.2	2.18
14	61	60.53	59	58.83	33	33.75	1.6	2.734
15	60	59.34	54	54.68	31	30.41	1.1	1.121
16	95	95.2	82	82	61	61.2	2.1	2.18
17	65	60.56	59	58.71	35	35.83	1.6	2.679
18	45	44.84	52	51.68	33	33.75	1.4	1.355
19	66	70.38	52	51.55	29	27.83	1.1	2.197
20	65	65.34	53	52.34	26	27.09	1.1	1.105
21	54	54.35	49	48.83	21	21.75	1.1	2.234
22	55	59.56	51	51.71	25	26.17	1	2.163
23	95	95.2	83	82	62	61.2	2.2	2.18
24	96	95.2	83	82	62	61.2	2.2	2.18
25	54	54.81	62	61.63	36	36.57	1.6	1.596
26	48	48.31	54	53.63	32	30.91	1.3	1.28
27	95	95.2	81	82	61	61.2	2.2	2.18
28	59	58.63	55	55.13	30	28.91	1.2	1.18
29	56	56.31	48	47.63	28	26.09	1.1	2.209

Experimental values - EV; Predicted values - PV



**Figure 1** The actual and predicted plot for (a) Methylparathion removal (%), (b) COD removal (%), (c) TOC removal (%) and (d) growth (OD)

#### Analysis of optimized process variables by response surface plots

The optimum values of the selected variables were obtained by solving their regression equation and analyzing response surface contour plots. Response Surface plots as a function of four factor at a time maintaining all other factors at a fixed level (zero for instance) are more helpful in understanding both the main and interaction effects of the four factors. The plots can be easily obtained by calculating the data from the model. The values were taken by one factor, where the second varies with constant of a given Y-values. The yield values of the different concentrations and range of the variable can also be predicted from respective response surface plots. The coordinates of the central point within the highest optimum concentration of the respective components. Figures 2 to 5 show their response surface obtained as a function of temperature, pH, time and agitation against methylparathion removal, COD removal, TOC removal and growth of bacteria in terms of optical density.

#### Optimum values and validation of the model

The methylparathion removal by *Pseudomonas aeruginosa* mpd was predominantly influenced by the combined effects of the environmental factors include temperature, pH, incubation period (time) and agitation. The point prediction from the analysis of variables for the response surface model showed the maximum methylparathion removal (95.2 %), COD removal (82 %), TOC removal (61.2 %) and growth (2.18 OD) by *Pseudomonas aeruginosa* (mpd) in

synthetic wastewater containing 1000 mgL<sup>-1</sup> of methylparathion at optimum conditions of pH (7), temperature (32.5 °C) and agitation at 150 rpm for 96 h of incubation period. As can be seen, there is not much difference between the experimental values and model response values obtained. This confirmed that RSM could be effectively used to predict the removal performance of methylparathion from wastewater by potential bacterial strain (*Pseudomonas aeruginosa* mpd).

The maximum experimental response for methylparathion removal was 95 % whereas the predicted value was 95.2 % indicating a strong agreement between them. The optimum values of the tested variables are at pH 7, 32.5°C temperature and agitation at 150 rpm for 96 h of incubation time as shown in perturbation graph (Figure 6.). The model was also validated by conducting the experiments under the optimized conditions, which resulted in the methylparathion removal of 96 % (Predicted response 95.2 %), thus proving the validity of the model.

The temperature is the most suitable variable for the growth of the isolates as well as the methylparathion (MP) removal which was found to be growth related processes. Temperature is another abiotic factor that influences the rate and extent of bioremediation since it affects microbial activity with rates of metabolic reactions generally increasing with increasing temperature (Baker, 1994 and Hong et al., 2007). Shake culture or aerated culture conditions are better for the growth and removal of methylparathion. The rise in temperature of the synthetic wastewater medium may accelerate the chemical reactions, reduces solubility of gases, amplifies taste and odour and elevates metabolic activity of organisms. This in turns reduce the organic loads in terms of COD and TOC in the wastewater. The decrease in COD and TOC may increase the biodegradation of methylparathion. This may be the cause for the increase in the biodegradability of methylparathion from the medium. So, the organic loads in terms of these parameters may increase the removal of methylparathion from aqueous solution. It was noted that the removal of organic load in terms of COD was proportional to the disappearance of cypermethrin (Jilani and Khan, 2006). Similar correlations were also observed by Berchtold et al., (1995).

The optimum growth of the strain was found to be pH 7. The strain mpd can degrade methylparathion from pH7; this could perhaps be due to increased bioavailability of methylparathion and optimal biotic activity of cells in this pH. The pH from 5 to 8 suggested that dissipation of methylparathion was mediated by the cometabolic activities of the bacteria and also the rate of degradation of methylparathion was low in acidic but increased considerably with an increase in pH. Brajesh et al., (2004) also reported similarly that the pH from 4.7 to 8.4 for chlorpyrifos by *Enterobacter* strain (B-14). The optimum conditions were more favorable for the growth of the bacteria. It may be either metabolize or co-metabolize the methylarathion in the medium as a nutrient or energy source for their growth and metabolism. The degradation of methylparathion supported cell growth, indicating that isolated strain could utilize methylparathion as a phosphorus source. The pH condition would be significance while emergent an effective remediation strategy.

The optimum time for the incubation period was enhanced the growth of the bacteria and increase its metabolic activity. The log phase of the bacteria was extended and the secondary metabolites which include the release of the respective enzymes responsible for the hydrolysis of methylparathion degradation, or oxidation and reduction process may occur. This in turns may results in the higher reactivity of the pollutant and increases the degradation process. The optimum agitation observed was more encouraged for the growth of the bacteria by utilizing the nutrients from the uniformly distributed and suspended nutrients in the medium which may helps in oxidation process. It may be either metabolize or co-metabolize the methylarathion in the medium. Shake culture or aerated culture conditions are better for the growth and removal of methylparathion. Methylparathion removal under aerobic conditions suggesting that a constitutively expressed enzyme could be involved in the degradation. Repeated application of pesticides results in the enhanced ability of microbial population to degrade the pesticide. The study also suggests that methylparathion degrading bacterial culture should preferably be used for the management of methylparathion containing wastewater.

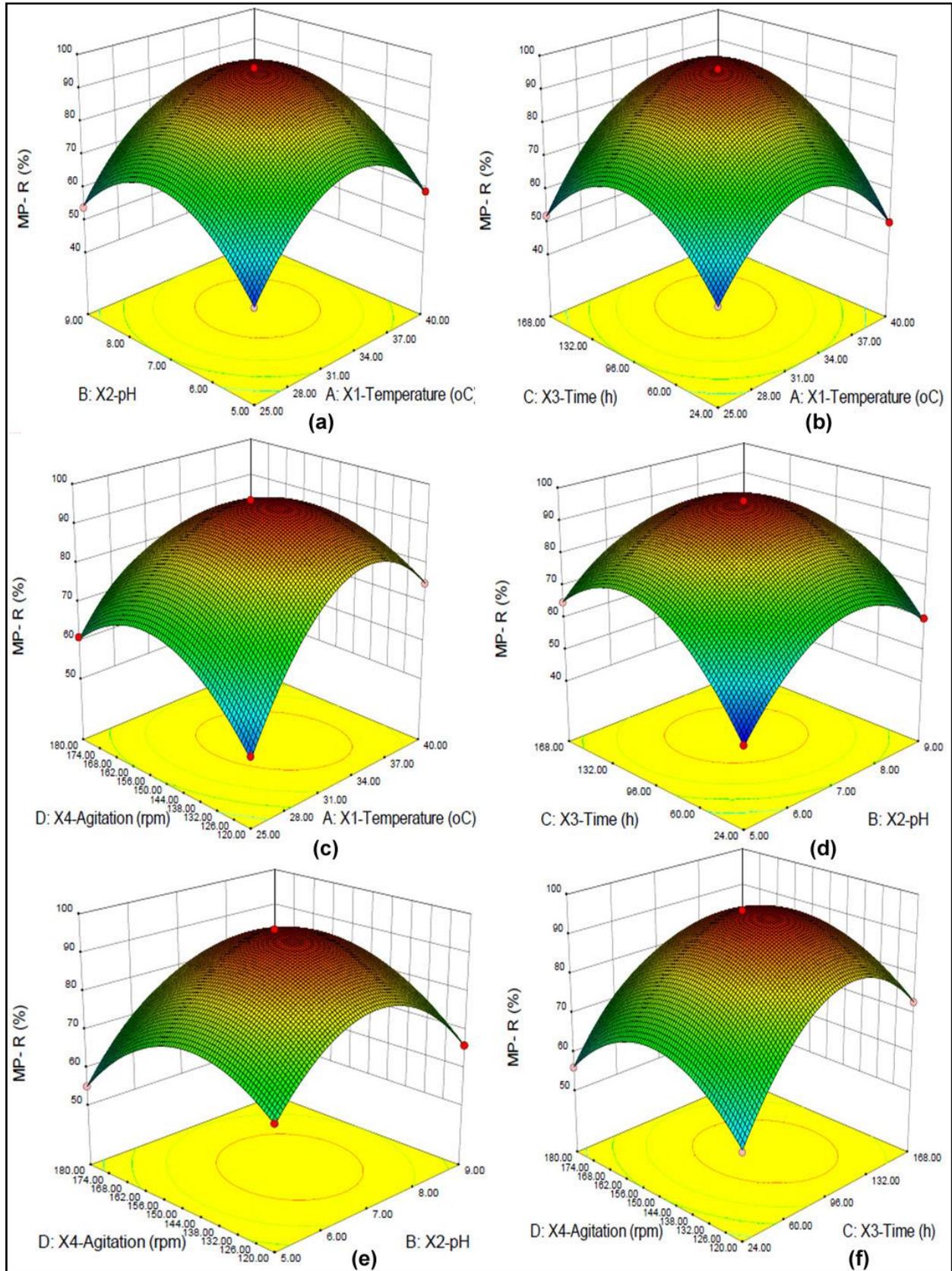
**Table 4** ANOVA table for Y<sub>1</sub> (methylparathion removal in %), Y<sub>2</sub> (COD removal in %), Y<sub>3</sub> (TOC removal in %) and Y<sub>4</sub> (growth in optical density) responses

Source	DF	Y <sub>1</sub>				Y <sub>2</sub>				Y <sub>3</sub>				Y <sub>4</sub>			
		SS	MS	F	P	SS	MS	F	P	SS	MS	F	P	SS	MS	F	P
<b>Model</b>	14	6903	493	1857	< 0.0001	3961	282	522	< 0.0001	4683	334	202	< 0.0001	4.55	0.32	117	< 0.0001
<b>A-X<sub>1</sub>- Temp (°C)</b>	1	261	261	984	< 0.0001	6.7	6.7	12.4	0.0033	18.7	18.7	11.3	0.0046	0.06	0.06	24.3	0.0002
<b>B-X<sub>2</sub>- pH</b>	1	90.7	90.7	341	< 0.0001	75	75	138	< 0.0001	80	80	48.4	< 0.0001	0.21	0.21	76.9	< 0.0001
<b>C-X<sub>3</sub>- Time (h)</b>	1	396	396	1494	< 0.0001	21.3	21.3	39.3	< 0.0001	10	10	6.1	0.0270	0.18	0.18	67.5	< 0.0001
<b>D- X<sub>4</sub>- Agitation (rpm)</b>	1	85.3	85.3	321	< 0.0001	80	80	147	< 0.0001	36.7	36.7	22.2	0.0003	0.16	0.16	58.8	< 0.0001
<b>X<sub>1</sub> X<sub>2</sub></b>	1	1	1	3.76	0.0727	9	9	16.6	0.0011	0.25	0.25	0.15	0.7032	0.00	0.00	0.90	0.3585
<b>X<sub>1</sub> X<sub>3</sub></b>	1	72.2	72.2	272	< 0.0001	4	4	7.38	0.0167	20.2	20.2	12.2	0.0035	0.00	0.00	0.90	0.3585
<b>X<sub>1</sub> X<sub>4</sub></b>	1	240	240	904	< 0.0001	72.2	72.2	133	< 0.0001	90.2	90.2	54.6	< 0.0001	0.12	0.12	44.1	< 0.0001
<b>X<sub>2</sub> X<sub>3</sub></b>	1	81	81	305	< 0.0001	4	4	7.38	0.0167	72.2	72.2	43.7	< 0.0001	0.25	0.25	90.1	< 0.0001
<b>X<sub>2</sub> X<sub>4</sub></b>	1	20.2	20.2	76.2	< 0.0001	4	4	7.38	0.0167	20.2	20.2	12.2	0.0035	0.06	0.06	22.5	0.0003
<b>X<sub>3</sub> X<sub>4</sub></b>	1	36	36	135	< 0.0001	25	25	46.1	< 0.0001	6.25	6.25	3.7	0.0722	0.09	0.09	32.4	< 0.0001
<b>X<sub>1</sub><sup>2</sup></b>	1	2808	2808	10579	< 0.0001	1138	1138	2102	< 0.0001	1655	1655	1001	< 0.0001	0.92	0.92	333	< 0.0001
<b>X<sub>2</sub><sup>2</sup></b>	1	2144	2144	8078	< 0.0001	914	914	1688	< 0.0001	1050	1050	635	< 0.0001	1.25	1.25	452	< 0.0001
<b>X<sub>3</sub><sup>2</sup></b>	1	2387	2387	8991	< 0.0001	1583	1583	2923	< 0.0001	1553	1553	940	< 0.0001	0.92	0.92	333	< 0.0001
<b>X<sub>4</sub><sup>2</sup></b>	1	1327	1327	5002	< 0.0001	1986	1986	3667	< 0.0001	2397	2397	1450	< 0.0001	2.07	2.07	746	< 0.0001
<b>Residual</b>	14	3.71	0.26			7.58	0.54			23.1	1.67			0.03	0.00		
<b>Lack of Fit</b>	10	2.91	0.29	1.458	0.3821	3.58	0.36	0.36	0.9143	20.3	2.0	2.9	0.157	0.03	0.00	1.54	0.3594
<b>Pure Error</b>	4	0.8	0.2			4	1			2.8	0.7			0.00	0.00		
<b>Cor Total</b>	28	6906				3968				4706				4.59			

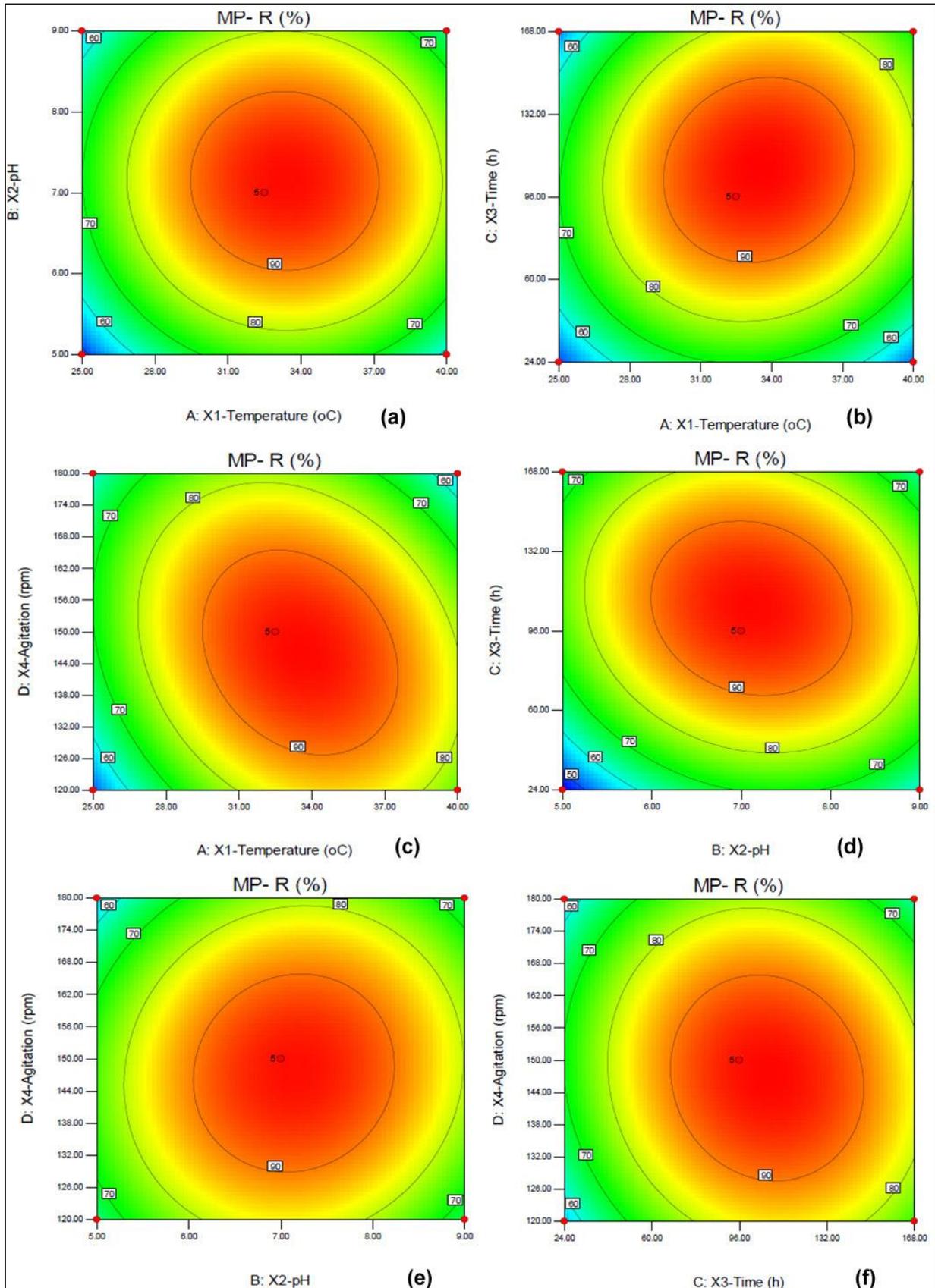
**Table 5** Analysis of variance (A NOVA) results of the model of methylparathion removal, COD and TOC removal by *Pseudomonas aeruginosa* mpd

Source	Sum squares	of Degree of freedom	of Mean Square	F-value	Prob > F	Remarks
<b><sup>a</sup>MP-Removal (%)</b>						
Model	6903	14	493.07	1857.3	<0.0001	<b>Significant</b>
Residual	3.716	14	0.2654			
Lack of fit	2.916	10	0.2916	1.4583	0.3821	Not- Significant
Pure error	0.8	4	0.2			
Cor Total	6906	28				
<b><sup>b</sup>COD Removal (%)</b>						
Model	3961	14	282.94	522.3	<0.0001	<b>Significant</b>
Residual	7.58	14	0.54			
Lack of fit	3.58	10	0.36	0.36	0.9143	Not- Significant
Pure error	4.00	4	1.00			
Cor Total	3968	28				
<b><sup>c</sup>TOC Removal (%)</b>						
Model	4683	14	334.55	202.47	<0.0001	<b>Significant</b>
Residual	23.13	14	1.6523			
Lack of fit	20.33	10	2.0333	2.9047	0.1579	Not- Significant
Pure error	2.8	4	0.7			
Cor Total	4706	28				
<b><sup>d</sup>G-MP<sup>+</sup> (OD)</b>						
Model	4.553	14	0.3252	117.25	<0.0001	<b>Significant</b>
Residual	0.038	14	0.0027			
Lack of fit	0.030	10	0.0030	1.5416	0.3594	Not- Significant
Pure error	0.008	4	0.002			
Cor Total	4.592	28				
<b>R-squared</b>	<b>Adj R-squared</b>	<b>Pred R-squared</b>	<b>Adequate precision</b>			
<sup>a</sup> R <sup>2</sup> = 0.9994;	R <sup>2</sup> <sub>adj</sub> = 0.9989;	R <sup>2</sup> <sub>pred</sub> = 0.9973	Adeq precision = 135.92			
<sup>b</sup> R <sup>2</sup> = 0.9981;	R <sup>2</sup> <sub>adj</sub> = 0.9962;	R <sup>2</sup> <sub>pred</sub> = 0.9932	Adeq precision = 69.66			
<sup>c</sup> R <sup>2</sup> = 0.9950;	R <sup>2</sup> <sub>adj</sub> = 0.9901;	R <sup>2</sup> <sub>pred</sub> = 0.9741	Adeq precision = 43.75			
<sup>d</sup> R <sup>2</sup> = 0.9915;	R <sup>2</sup> <sub>adj</sub> = 0.9830;	R <sup>2</sup> <sub>pred</sub> = 0.9586	Adeq precision = 30.60			

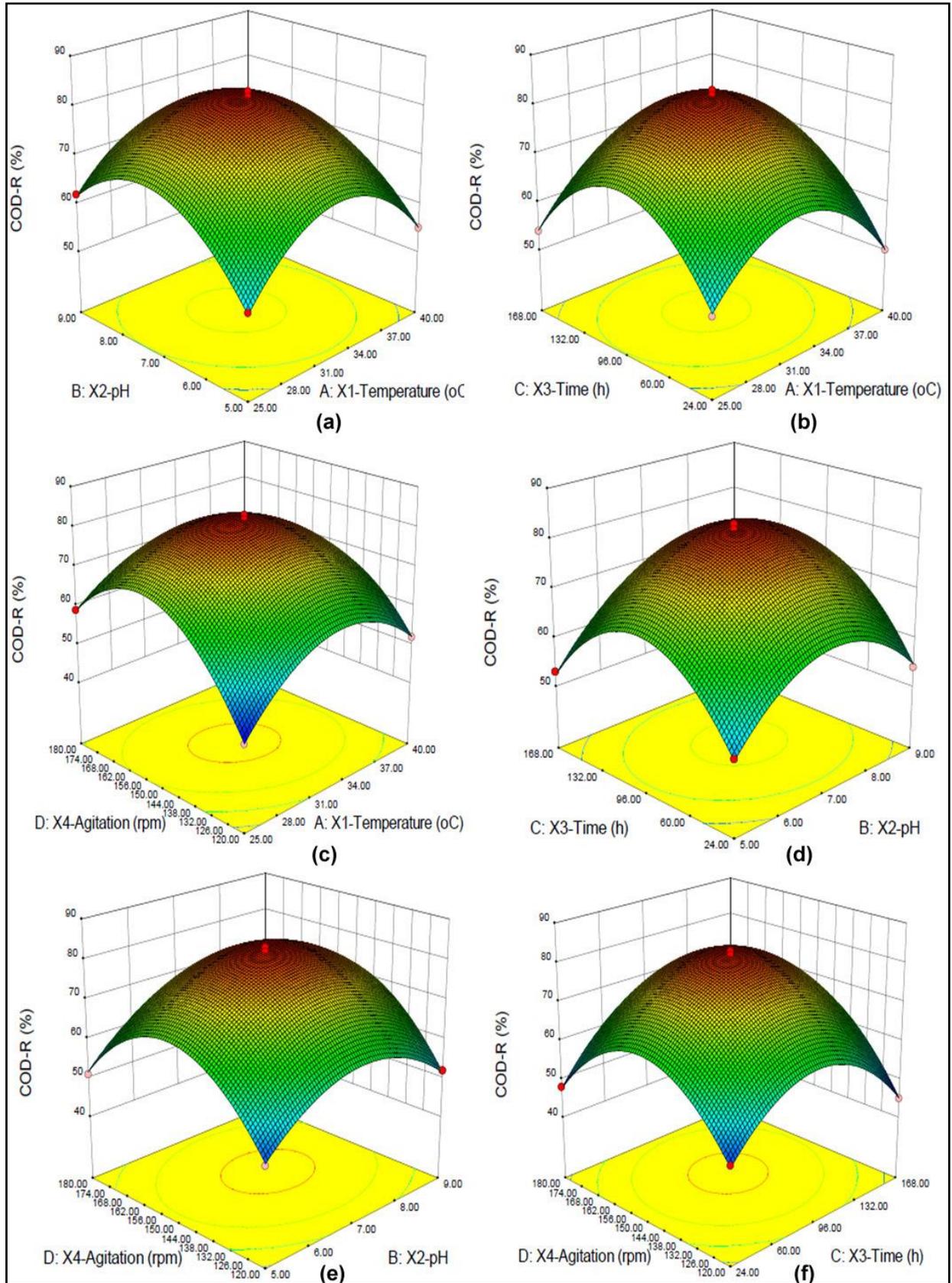
*MP- Methylparathion, COD- Chemical oxygen demand, TOC- Total organic carbon, G-MP<sup>+</sup> - Growth in presence of methylparathion, OD- Optical density*



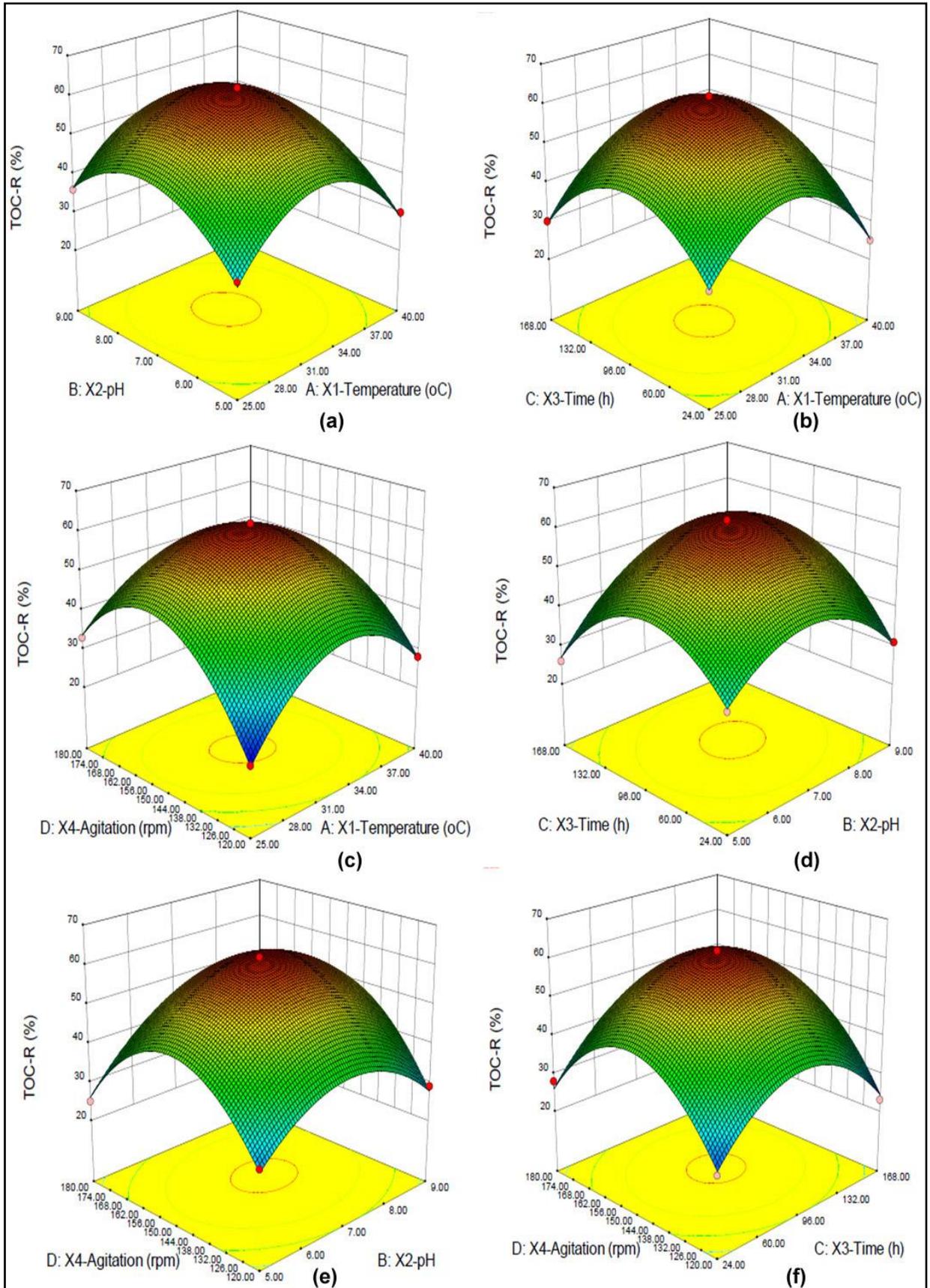
**Figure 2a** Response surface plot of the combined effects of pH, temperature, time and agitation on the percentage removal of Methylparathion by *Pseudomonas aeruginosa* mpd



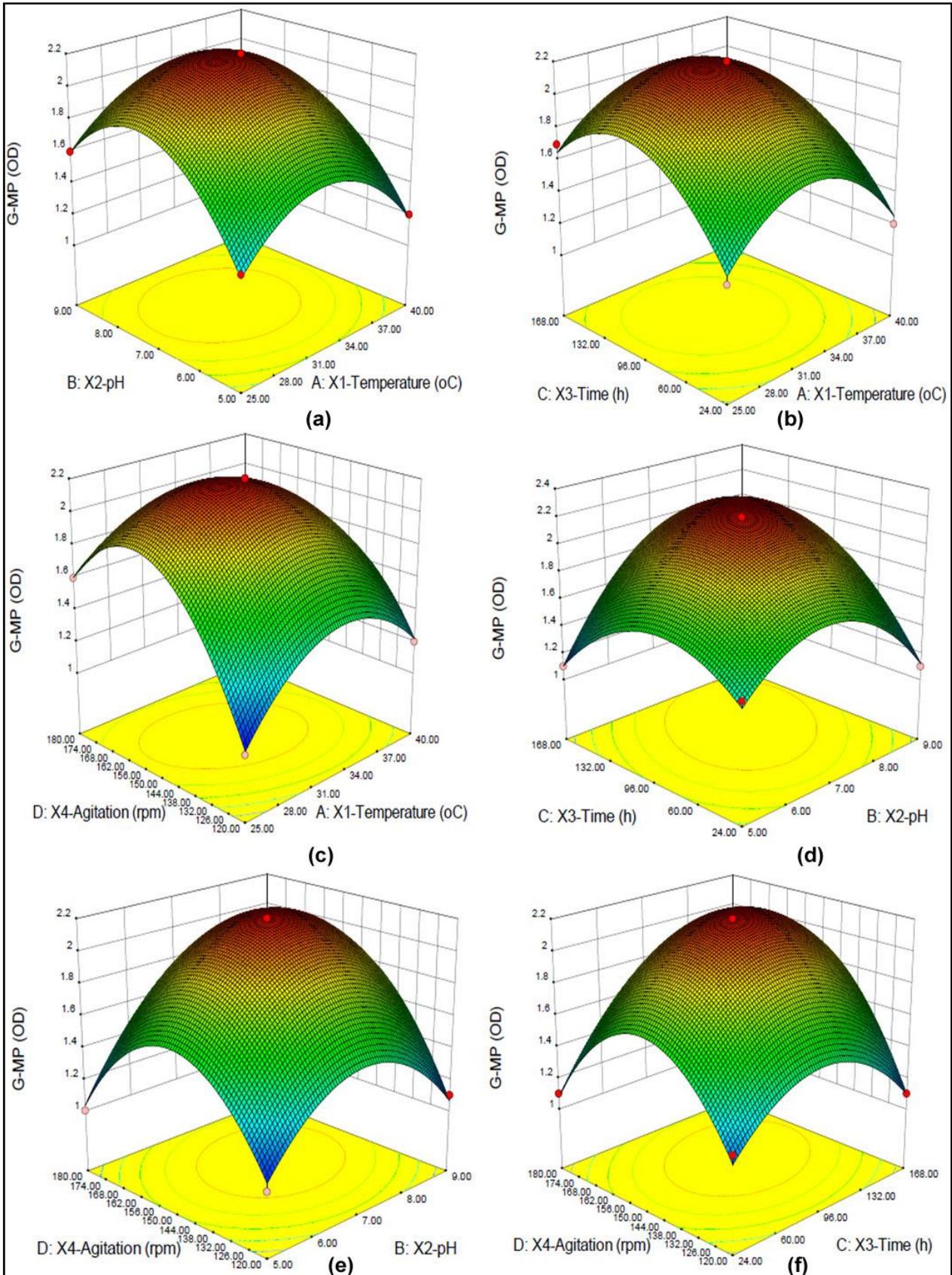
**Figure 2b** Contour surface plot of the combined effects of pH, temperature, time and agitation on the percentage removal of Methylparathion by *Pseudomonas aeruginosa* mpd



**Figure 3** Response surface plot of the combined effects of pH, temperature, time and agitation on the percentage removal of COD by *Pseudomonas aeruginosa*



**Figure 4** Response surface plot of the combined effects of pH, temperature, time and agitation on the percentage removal of TOC by *Pseudomonas aeruginosa* mpd



**Figure 5** Response surface plot of the combined effects of pH, temperature, time and agitation on the growth of *Pseudomonas aeruginosa* mpd

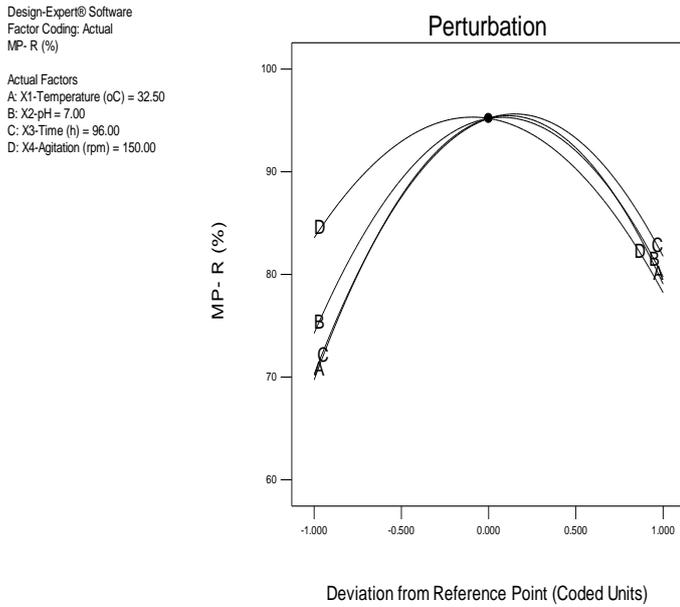


Figure 6 Perturbation graph showing the optimum values of the tested variables

However, in the presence of other carbon sources (such as glucose), initially it delayed to degrade methylparathion but with the passage of time it degraded to 95 % within 96 h, indicating that when glucose was depleted, it started to utilize methylparathion as a source of carbon. Similar results were reported by **Brajesh et al., (2004)**. Glucose was chosen because it is a primarily substratum and the main carbon source for the bacteria. Glucose addition is important to improve the efficacy of bioremediation of persistent compounds like pesticides (**Sampaio, 2005; Singh, 2006; Yang et al., 2009; Yugui et al., 2008**). **Qiu et al., (2006)** reported that the additional nutrients such as glucose and organic nitrogen greatly enhanced the growth of *Ochrobactrum* sp B2. **Singh (2006)**, reports that the addition of glucose produces substances of high reactivity, which react more easily with the pollutant. Previous reports concerning isolation of organophosphorus degrading microorganisms suggest that the bacteria mainly degrade the compounds cometabolically (**Horne et al., 2002; Zhongli et al., 2001**). Some reports showed that the isolated bacterium can utilize organophosphates as a source of carbon or phosphorus (**Subhas and Dileep, 2003**) from the hydrolysis products (**Serdar and Gibson, 1985**). In natural environments, the competition for carbon sources is immense and the utilization of pesticide as an energy source by this bacterium provides it with a substantial competitive advantage over other microorganisms (**Malghani et al., 2009**).

UV-Vis Spectral Analysis

In order to investigate the formation and eventual disappearance of intermediate compounds in the reaction mixture, the biotreated synthetic wastewater containing methylparathion was monitored using UV-Vis spectroscopy as a function of time. The UV-Vis spectroscopy scanning profile shows a peak formation with lambda max ( $\lambda_{max}$ ) at 277 nm as shown in Figure 7. The extended biotreatment after 96 h shows the same band decrease its intensity and eventually disappeared. The absorbance value was found to be reduced at maximum time of 96 h at optimized process variables. The wavelength at 277nm shows a displacement to higher wavelengths and formation of band at 400 nm that can be attributed to the p-nitrophenol absorption bands. **Zhongli et al (2001)** reported that the maximum absorption peak of methylparathion was recorded at 273nm by *Plesiomonas* strain (M6). **Wu and Linden (2008)** reported that the parathion produces a maximum absorbance ( $\lambda_{max}$ ) at 275nm. Further, the biotreated samples were analysed by HPLC for the confirmation of the residual MP and intermediates formation.

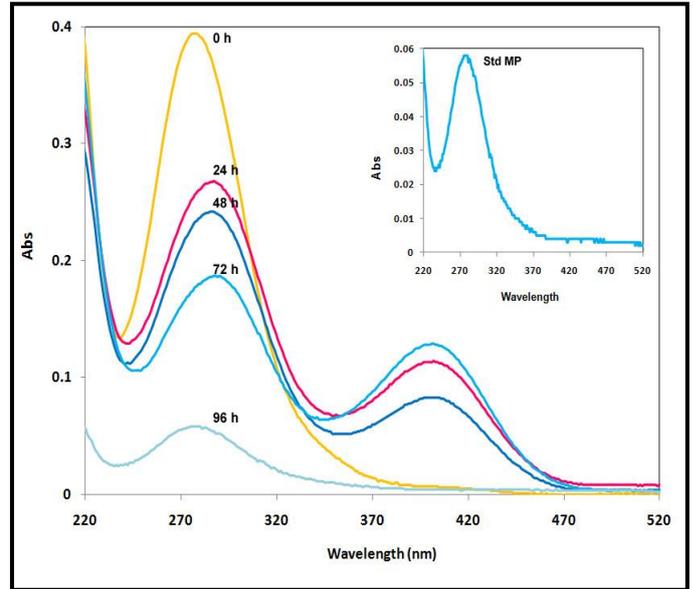


Figure 7 UV –Vis NIR Spectroscopic scanning profile of MP degradation by *Pseudomonas aeruginosa* mpd at different treatment time (h)

HPLC Analysis

The biotreated samples were analysed by HPLC for the confirmation of the residual methylparathion and their byproducts or intermediates formation. The retention time for methylparathion was found to be 3.3min which was confirmed by the spectra as shown in Figure 8. The percentage degradation of methylparathion by *Pseudomonas aeruginosa* mpd was found to be 95 %. Treated samples showed that the peak reduction at 3.3 retention time (RT), hence it proves the degradation of methylparathion by biotreatment (*Pseudomonas aeruginosa* mpd) process. The peak at retention time of 4.0, 4.4 and 10 min in treated sample were observed as the intermediate products of methylparathion degradation during the biotreatment process. Moreover, methylparathion was rapidly oxidized into other organic compounds.

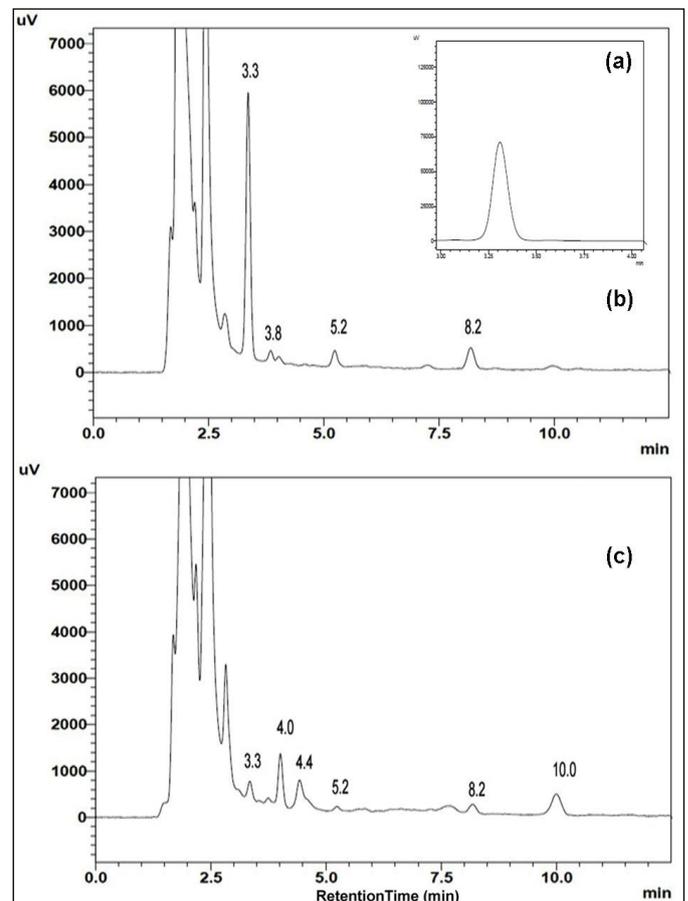


Figure 8 HPLC profiles of methylparathion biodegradation by *Pseudomonas aeruginosa* mpd after 72h (a) Standard, (b) Control and (b) Treated after 72 h

## CONCLUSION

RSM was used in this study to establish the optimum variables initial pH, time, culture temperature and agitation for methylparathion biodegradation. It was concluded that the optimal conditions for methylparathion removal are pH 7 and 32.5 °C temperature and agitation at 150 rpm for 96 h of incubation period. The predicted extent of methylparathion biodegradation by this strain of *Pseudomonas aeruginosa* under these optimum conditions was 95.2 %, and the experimental results were in close agreement with this prediction. The point prediction from the analysis of variable for response surface model for methylparathion removal (95.2 %), COD removal (82 %), (c) TOC removal (61.2 %) and (d) growth (2.18 OD) by *Pseudomonas aeruginosa* (mpd) from waste water with 1000 mg/L of medium at optimum conditions of pH, temperature and agitation for 96 h of incubation period. The predicted optimal and experimental measured methylparathion removal efficiencies agreed well with high coefficients of determination ( $R^2 = 0.9994$ ,  $R^2_{adj} = 0.9989$ ), and the COD removal ( $R^2 = 0.9981$ ,  $R^2_{adj} = 0.9962$ ) and TOC removal ( $R^2 = 0.9950$ ,  $R^2_{adj} = 0.9901$ ) are also agreed well. Moreover the growth of the strain in terms of its OD were also agreed well ( $R^2 = 0.9915$ ,  $R^2_{adj} = 0.9830$ ). Hence this study was an attempt for methylparathion removal using *Pseudomonas aeruginosa* strain with RSM model, has helped to recognize the important operating variables and optimum levels with least effort and time. The isolate of the present study was found to have potential in methylparathion removal at optimized condition and suggested for biotreatment of methylparathion wastewater. This study will form the basis for the further utilization of the bacterial strain, grown on suitable substrates, in biofiltration systems for the treatment of wastewaters.

**Acknowledgments:** The author Ms. K. Usharani expresses her sincere thanks to the editor and anonymous reviewers for present the research paper in a professional manner.

**Conflict of interest:** The authors declare no conflict of interest.

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